# PRENYLATED PHTHALIDES FROM ANAPHALIS ARANEOSA AND HELICHRYSUM PLATYPTERUM

J. JAKUPOVIC, A. SCHUSTER, H. SUN, F. BOHLMANN and D. S. BHAKUNI\*

Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, F.R.G.; \*Central Drug Research Institute, Lucknow, India

(Received 14 May 1986)

Key Word Index—Anaphalis araneosa; Helichrysum platypterum; Compositae; phthalides.

Abstract—The aerial parts of Anaphalis araneosa afforded, in addition to known compounds, three new prenylated phthalides. A further compound of this type was isolated from the roots of Helichrysum platypterum. The structures were elucidated by high-field <sup>1</sup>H NMR and <sup>13</sup>C NMR spectroscopy.

## INTRODUCTION

A few species of the genus Anaphalis (Compositae, tribe Inuleae, subtribe Gnaphaliinae) with about 60-100 species [1] have been studied chemically. In addition to flavanoids [2, 3], helipyrone [4], typical chloroacetylenic compounds with a dihydropyrane or furane ring [5] and two phthalides were reported [6]. Also A. araneosa DC. has been studied chemically [2, 3]. We have now reinvestigated this species and also the roots of Helichrysum platypterum. The results are discussed in this paper.

## RESULTS AND DISCUSSION

The aerial parts of A. araneosa afforded 5,7-dimethoxyflavone, loliolide, sitosterol, stigmasterol and the phthalides 1, 2 and 4. The structure of 1 followed from the molecular formula (C14H14O4) and the 1H NMR spectral data (see Experimental). The presence of a 2,2-dimethylchromene derivative was deduced from the typical signals at  $\delta$ 1.48 s (6H), 6.16 and 5.61 d (each 1H, J = 10 Hz). The presence of an aromatic signal at  $\delta 6.36 \text{ s}$  (1H), together with signals at  $\delta 5.16$  s (2H) and 3.93 s (OMe), indicated the presence of three further substituents, one being a methoxyl group while the two remaining must form a  $\gamma$ -lactone ring, as followed from the molecular formula and the IR spectrum. The relative positions of the substituents were determined by NOE difference spectroscopy. Clear effects were obtained between H-3 and H-10 as well as between the methoxy protons and H-6. We have named compound 1 phthalidochromene.

The <sup>1</sup>H NMR spectrum of 2 (see Experimental) indicated the presence of a dihydrofurane derivative combined with a phthalide. Comparison with the data of related compounds showed that the dihydrofurane was substituted with a methoxyl group which was in the *trans*-position to a hydroxyisopropyl group. Accordingly, the proposed structure for 2, which we have named araneophthalide, was very likely.

Compound 4 showed an unclear <sup>1</sup>H NMR spectrum with highly broadened overlapped multiplets. However, after acetylation, a clear spectrum was obtained which showed the typical signals of a tetraacetate of a  $\beta$ -glucopyranoside (see Experimental). Singlets at  $\delta 2.73$  (2H), 1.49 (3H) and 1.50 (3H) indicated the presence of a

dimethyldihydrochromanone. The signal from the only aromatic proton was shifted downfield when compared with the chemical shift of H-6 in the spectrum of 1, and the signal of H-3 was replaced by a pair of doublets at  $\delta$ 5.46 and 5.41. All data therefore agreed well with the proposed structure of 4a, which was supported by the <sup>13</sup>C NMR spectrum (see Experimental). The <sup>13</sup>C NMR spectrum of the free glucoside 4 also showed broadened signals and therefore was much less useful. We have named the glucoside 4 aranochromanophthalide.

From the roots of Helichrysum platypterum DC. the phthalide 3 was isolated. The presence of a free hydroxyl group was shown by the formation of the methyl ether 3a. The <sup>1</sup>H NMR spectra of 3 and 3a (see Experimental) were close to that of 2. However, the typical signals of an

Short Reports

isopropenyl group, a pair of double doublets and a broadened triplet indicated a dihydrobenzofurane derivative. All data therefore agreed with the proposed structure, which was further supported by the <sup>13</sup>C NMR spectrum of 3, which we have named platypterophthalide.

The isolation of further phthalides may be of chemotaxonomic interest. So far, these relatively rare types of natural products have also been isolated from Helichrysum species [7-10] and from Anaphalis contorta [6]. Especially closely related to 14 are the prenylated phthalides from Helichrysum arenarium [7] and from the Anaphalis species [6]. The unusual polyynes with a dihydropyrane moiety are also present in several Helichrysum and Anaphalis species [5]. Further studies may show whether the prenylated phthalides are more widespread in the subtribe Gnaphaliinae.

### **EXPERIMENTAL**

The air-dried plant material was extracted with Et<sub>2</sub>O-MeOH-petrol (1:1:1) and the extracts obtained were separated as reported previously [11]. The extract of the aerial parts of Anaphalis araneosa (voucher 2729, deposited at the Herbarium of the Central Drug Research Institute of Lucknow, India, 10.36 g) was first separated by CC (SiO<sub>2</sub>) into four fractions (1: Et<sub>2</sub>O-petrol, 1:1; 2: Et<sub>2</sub>O; 3: Et<sub>2</sub>O-MeOH, 9:1; and 4: Et<sub>2</sub>O-MeOH, 3:1). Prep. TLC of fraction 1 (Et<sub>2</sub>O-petrol, 1:3) gave 50 mg sitosterol, 50 mg stigmasterol and 20 mg 5,7-dimethoxyflavone. HPLC (RP 8, MeOH-H<sub>2</sub>O, 7:3, ca 100 bar) of fraction 2 gave two crude fractions (2/1 and 2/2). Prep. TLC (CHCl<sub>3</sub>-MeOH, 20:1) of fraction 2/1 gave 5 mg loliolide and fraction 2/2 3.8 mg 1. HPLC of fraction 3 (MeOH-H<sub>2</sub>O, 13:7) gave 2 mg 2 and HPLC (MeOH-H<sub>2</sub>O, 1:1) of fraction 4 afforded 18 mg 4.

The extract of the roots (210 g) of Helichrysum platypterum (voucher 81/267, deposited at the National Botanic Research Institute, Pretoria, Republic of South Africa) gave on CC (silica gel) and prep TLC (Et<sub>2</sub>O-petrol, 3:1) 2 mg 3.

Phthalidochromene (1). Colourless gum; IR  $v_{\text{max}}^{\text{CCI}}$  cm<sup>-1</sup>: 1760 ( $\gamma$ -lactone); MS m/z (rel. int.): 246.089 [M]<sup>+</sup> (14) (calc. for C<sub>14</sub>H<sub>14</sub>O<sub>4</sub>: 246.089), 231 [M - Me]<sup>+</sup> (100), 203 [231 - CO]<sup>+</sup> (8); CIMS m/z (rel. int.): 247 [M + 1]<sup>+</sup> (100); <sup>1</sup>H NMR (CDCl<sub>3</sub>, always 400 MHz):  $\delta$ 5.16 (s, H-3), 6.36 (s, H-6), 6.16 (d, H-10, J = 10 Hz), 5.61 (d, H-11, J = 10 Hz), 1.48 (s, 6H, H-13, H-14), 3.93 (s, OMe).

Araneophthalide (2). Colourless gum;  $IR v_{max}^{CCL} cm^{-1}$ : 3600 (OH), 1760 (y-lactone); MS m/z (rel. int.): 294.110 [M]<sup>+</sup> (1.3) (calc. for  $C_{15}H_{18}O_6$ : 294.110), 262 [M - MeOH]<sup>+</sup> (4), 244 [262 -  $H_2O$ ]<sup>+</sup> (24), 111 (100); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 5.23 and 5.17 (d, H-3, J = 16 Hz), 6.44 (s, H-6), 5.06 (d, H-10, J = 3 Hz), 4.53 (d, H-11, J = 3 Hz), 1.33 (s, H-13), 1.32 (s, H-14), 3.96 and 3.35 (s, OMe).

Platypterophthalide (3). Colourless gum; IR  $\nu_{\text{max}}^{\text{CCL}}$  cm<sup>-1</sup>; 3590 (OH), 1760 (γ-lactone); MS m/z (rel. int.): 232.074 [M]<sup>+</sup> (72) (calc. for C<sub>13</sub>H<sub>12</sub>O<sub>4</sub>: 232.074), 217 [M – Me]<sup>+</sup> (100); CIMS m/z (rel.

int.): 233 [M + 1]<sup>+</sup> (100); <sup>1</sup> H NMR (CDCl<sub>3</sub>): 5.20 (s, H-3), 6.40 (s, H-6), 3.35 (dd, H-10, J = 16, 9.5 Hz), 3.00 (dd, H-10', J = 16, 8 Hz), 5.34 (dd, H-11, J = 8, 9.5 Hz), 5.08 and 4.94 [s (br), H-13], 1.76 [s (br), H-14], 7.76 (s, OH); <sup>13</sup>C NMR (C-1 and C-3-C-14):  $\delta$ 168.4 s, 70.2 t, 112.7 s, 153.0 s, 95.8 d, 164.4 s, 104.6 s, 143.0 s, 30.4 t, 88.3 d, 149.0 s, 112.8 t, 17.0 q.

581

Addition of CH<sub>2</sub>N<sub>2</sub> in Et<sub>2</sub>O gave 3a, colourless gum; MS m/z (rel. int.): 246 [M]<sup>+</sup> (100), 231 [M - Me]<sup>+</sup> (81); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 5.13 (s, H-3), 3.48 (dd, H-10, J = 16, 9.5 Hz), 3.11 (dd, H-10', J = 16, 8 Hz), 5.30 (dd, H-11, J = 8, 9.5 Hz), 5.09 and 4.95 [s (br), H-13], 1.77 [s (br), H-14], 4.16 (s, OMe).

Aranochromanophthalide (4). Amorphous, colourless material; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 60°): δ5.28 [s (br), H-3], 6.58 [s (br), H-6], 2.70 and 2.64 [d (br), H-11], 1.46 [s (br), H-13], 1.44 [s (br), H-14], 5.09 [d (br), H-1'], 3.9-3.4 (m, H-2'-H-6');  $^{13}$ C NMR (CDCl<sub>3</sub>):  $\delta$ 190.6 s, 168.9 s, 166.9 s, 161.0 s, 152.3 s, 110.1 s, 105.1 d, 101.3 d, 81.7 s, 76.7 d, 76.1 d, 73.1 d, 72.3 d, 70.8 t, 69.9 d, 61.7 t, 48.6 t, 26.8 q, 26.6 q. 8 mg 4 in 1 ml Ac<sub>2</sub>O and 0.1 ml pyridine on standing for 12 hr gave, after usual work-up, 6 mg 4a, colourless gum; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  5.46 and 5.41 (d, H-3, J = 18 Hz), 6.70 (s, H-6), 2.73 (s, H-11), 1.50 (s, H-13), 1.49 (s, H-14), 5.25 (d, H-1', J = 7.5 Hz), 5.39 (dd, H-2', J = 7.5, 9.5 Hz), 5.31 (dd, H-3', J = 9, 9.5 Hz), 5.14 (dd, H-2', J = 9, 9.5 Hz), 5.14 (dd, H-2', J = 9, 9.5 Hz), 5.14 (dd, H-3', J = 9, 9.5 Hz), 6.14 (dd, H-3', J = 9, 9.5 Hz),H-4', J = 9, 9.5 Hz), 3.96 (ddd, H-5', J = 9.5, 6, 3 Hz), 4.26 (dd, H-5'), 4.27 (dd, H-5'), 4.28 (dd, H-5'), 4.2  $6_1'$ , J = 12, 3 Hz), 4.21 (dd, H- $6_2'$ , J = 12, 6 Hz); OAc:  $\delta 2.14, 2.08$ , 2.06 and 2.04 s; <sup>13</sup>C NMR (CDCl<sub>3</sub>, C-1 and C-3-C-14): 166.6 s, 70.1 t, 110.5 s, 160.1 s, 105.8 d, 166.2 s, 109.2 s, 152.3 s, 190.5 s, 48.3 t, 81.5 s, 26.6 q, 26.7 q; C-1'-C-6': 99.1 d, 70.3 d, 72.6 d, 68.3 d, 72.1 d, 62.0 t; OAc 170.4, 170.2, 169.3, 169.3 s, 20.6 q, 20.5 q (3  $\times$  ).

### REFERENCES

- Merxmüller, H., Leins, P. and Roessler, H. (1977) in The Biology and Chemistry of the Compositae (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds), p. 581. Academic Press. London.
- Ali, E., Bagchi, D. and Pakrashi, S. C. (1979) Phytochemistry 18, 356.
- Lin, J. H., Lin, Y. M. and Chen, F. C. (1976) J. Chin. Chem. Soc. 23, 57.
- 4. Ali, E., Bagchi, D. and Pakrashi, S. C. (1982) Phytochemistry
- Bohlmann, F., Burkhardt, T. and Zdero, C. (1973) Naturally Occurring Acetylenes, p. 352. Academic Press, London.
- Talapatra, B., Roy, M. K. and Talapatra, S. K. (1980) Planta Med. 39, 223.
- Vrkoc, J., Budesinsky, M., Dolejs, L. and Varickova, S. (1975) Phytochemistry 14, 1845.
- 8. Opitz, L. and Hänsel, R. (1971) Arch. Pharm. 304, 228.
- Zapesochnaya, G. G., Bankovskii, A. I. and Nakaidze, A. K. (1972) Khim. Prir. Soedin. 6, 804.
- Hänsel, R., Langhammer, L. and Albrecht, A. G. (1963) Sci. Pharm. 31, 88.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1984) Phytochemistry 23, 1979.